

bismuth

Russell

SEMI-CYCLIC ALTERATION

In an [electric current](#) there is a constant interchange between [anode](#) and [cathode](#) or [positive](#) and [negative](#) poles. A [light particle](#) expands as it leaves the [anode](#) in an outward radial direction and contracts as it radially approaches the [anode](#). This [light particle](#) has been the same [light particle](#) at all times in all parts of its journey. Its variation of [charge](#) and [discharge](#), its [direction](#) of [motion](#) and the [condition](#) of [wave pressure](#) in which it finds itself at any time are the sole reasons for its changing from one [condition](#) to another. The [light particles](#) are all the same [light particles](#), all being different only in [pressure condition](#).

This is also true of the elements of [matter](#). Whether they be [iron](#), [carbon](#), [silicon](#), **bismuth** or [radium](#), all are composed of the same kind of [light particles](#).

They all seem to have different qualities and attributes, but those qualities and attributes are likewise given to them purely by the positions they occupy in their [waves](#). [Walter Russell, [The Secret of Light](#), pages 167-168]

Schauberger

[16] List of [paramagnetic](#) and [diamagnetic](#) elements:

1. Apart from [iron](#), [nickel](#) and [cobalt](#), whose [magnetic properties](#) are already known, [osmium](#) and almost all [iron](#) compounds are [paramagnetic metals](#).
2. **Bismuth** and [antimony](#) are particularly [diamagnetic](#). [Zinc](#), [tin](#), [lead](#), [copper](#), [silver](#) and [gold](#) as well as [glass](#) and [carbon disulphide](#) and other [non-conductors](#) are strongly [diamagnetic](#). [Aloys Kokaly, [Implosion Magazine](#), No. 45, p. 19. For further elaboration of the various forms of [magnetism](#), see Chapter 2, endnote 23, p. 88, [The Fertile Earth](#), Vol. III of the [Ecotechnology series](#). - Ed.] [[The Energy Evolution - Harnessing Free Energy from Nature](#), [The Catalysts](#)]