

Quantum Transition

Discontinuous transitions of a [quantum](#) system (an atom, a [molecule](#), an atomic [nucleus](#), a solid) from one state to another. The most important are **quantum transitions** between stationary states that correspond to different energies of the quantum system - the **quantum transitions** of a system from one energy level to another. During a transition from a higher energy level - to a lower level the system releases the energy - and during the reverse transition, absorbs it. [The Great Soviet Encyclopedia (1979)], [Quantum Transitions](#) ↗

See Also

15.18 - Keelys Process for Liberating Ether from Water wherein Keely describes transitions from [molecular](#) to [atomic](#) to [etheric](#) ([plasma](#)) states.

Subdivision

Transformation

Transmutation