

# Deuteron



## Deuteron

([click to enlarge](#))

Structure:

[Neutron](#)

[Proton](#)

Characteristics:

According to the recommendations of IUPAC Commission on Physical Organic Chemistry [Pure Appl. Chem., 60, 1115-1116 (1988)], the names for [hydrogen atoms](#) and [ions](#) are the following:

	1H	2H	3H	H
atom	<a href="#">protium</a>	<a href="#">deuterium</a>	<a href="#">tritium</a>	<a href="#">hydrogen</a>
cation	<a href="#">proton</a>	<a href="#">deuteron</a>	<a href="#">triton</a>	<a href="#">hydron</a> (female)
anion	<a href="#">protide</a>	<a href="#">deuteride</a>	<a href="#">tritide</a>	<a href="#">hydride</a> (male)

## Schauberger

[1] See "The Ox and the Chamois" in [Nature as Teacher](#), p.41, Vol. II of the [Ecotechnology series](#). — Ed.

[2] [H-substance](#): here refers to [hydrogen](#) or **hydrogen-like substances**. — Ed.

[3] [Phos-elements](#): It is not quite clear what is intended here, but it may relate in some way to [bioluminescence](#). However, the following three definitions are provided as an aid to interpretation.

**PHOSPHOR**: A [substance](#) which is capable of [luminescence](#), i.e. storing [energy](#) (particularly from [ionising radiation](#)) and later releasing it in the form of [light](#). If the [energy](#) is released after only a short delay (between 10<sup>-10</sup> and 10<sup>-4</sup> seconds) the [substance](#) is called a '[scintillator](#)'.

**PHOSPHORUS**: P. Element. [Atomic weight](#) 30.9738. [Atomic number](#) 15. Occurs in several [allotropic forms](#), white phosphorus and red phosphorus being the commonest. The former is a waxy white, very inflammable and poisonous solid. Red phosphorus is a non-poisonous, dark red powder, not very inflammable. The element only occurs in the combined state, mainly as [calcium phosphate](#), CA3(PO4)2, Essential to [life](#); [calcium phosphate](#) is the main constituent of animal [bones](#).

**PHOSPHATE**: Salt of phosphoric acid H3PO4. Phosphates are used as [fertilisers](#) to rectify a deficiency of [phosphorus](#) in the [soil](#). Note: The editor regrets that he cannot locate the dictionary from which the information was originally sourced.' [[The Energy Evolution - Harnessing Free Energy from Nature](#), [Letter to Werner Zimmermann](#)]

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ChatGPT Compares hydrogen to SVP principles and laws [12/29/24]:

<https://chatgpt.com/share/6771397c-49b8-800d-840e-eae5de509212>

ChatGPT further compares hydrogen to SVP principles and laws 12/29/24: [1]

<https://chatgpt.com/share/6771397c-49b8-800d-840e-eae5de509212>

See Also

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## AI Interpretations of SVP Hydrogen

## Hydrogen - Snell

[NUC PHYS] The [nucleus](#) of a [deuterium atom](#), consisting of a [neutron](#) and a [proton](#). Designated d. Also known as [deuteron](#).

**Deuterium:** [deuterium](#) (a [hydrogen](#) isotope found as a natural component of seawater)

[CHEM] The isotope of the element [hydrogen](#) with one [neutron](#) and one [proton](#) in the [nucleus](#); [atomic weight](#) 2.0144. Designated D, d, H<sub>2</sub>, or 2H.

Heavy Water

In paper after paper last week, scientists reported that when a metal, usually [palladium](#), absorbs huge amounts of [deuterium](#) into its atomic lattice, the result is more [heat](#) than plain old electrochemistry can explain, as well as particles thought to be by-products of nuclear [fusion](#).

In the cell, heavy [hydrogen](#) is forced into the [palladium](#) until a new class of nuclear reactions occurs, in which energy of great [intensity](#) is released without the deadly [radiation](#) or radioactive by-products produced by other nuclear energy processes.

NOTE: [Keely](#) absorbed [Hydrogen](#) into '[platina](#)'.

See Also

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**Etheric Elements**  
**Particles and Corpuscles**  
**Table of Quantum Particles**